

Quantitative Analysis Solution Manual

Chapter 1 : Quantitative Analysis Solution Manual

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9 gas – solution equilibrium k_h henry's law co₂ dissolves in water: co₂ (g) + h₂o <=> h₂co₃ (aq) k_h = 3.4 x 10⁻² at a co₂ pressure of 3 x 10⁻⁴ atmospheres, what is the concentration of the carbonic acid in the water? 10⁻⁵ m water equilibrium kw

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